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P&G Case CM-2478

IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

In the application of :
Sanjeev Sharma, et al. : Confirmation No. 8739
Serial No. 09/994,557 : Group Art Unit 1744
Filed November 27, 2001 : Examiner Not Yet Assigned
For DISHWASHING METHOD

TRANSMITTAL OF CERTIFIED COPIES OF PRIORITY DOCUMENTS

Assistant Commissioner for Patents
Washington, D.C. 20231

Dear Sir:

Applicants hereby submit certified copies of priority documents for Patent Application No. 09/994,557, filed November 27, 2001.

Also being submitted is an Application Data Sheet.

Respectfully submitted,

By *Jeffrey W. Damber*
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Date: February 25, 2003

Customer No. 27752

(trans-priority.doc)
Last revised: 04/11/02

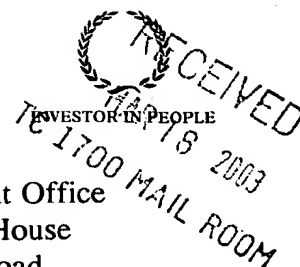
$$\begin{aligned}
 \mathcal{L}_1 &= \mathcal{L}_1^{\text{reg}} + \mathcal{L}_1^{\text{KL}} \\
 \mathcal{L}_1^{\text{reg}} &= \frac{1}{2} \text{tr}(\mathbf{C}) + \frac{1}{2} \text{tr}(\mathbf{C}^{-1}) \\
 \mathcal{L}_1^{\text{KL}} &= -\frac{1}{2} \log |\mathbf{C}| - \frac{1}{2} \text{tr}(\mathbf{C}^{-1}) \\
 \mathcal{L}_2 &= \mathcal{L}_2^{\text{reg}} + \mathcal{L}_2^{\text{KL}} \\
 \mathcal{L}_2^{\text{reg}} &= \frac{1}{2} \text{tr}(\mathbf{C}) + \frac{1}{2} \text{tr}(\mathbf{C}^{-1}) \\
 \mathcal{L}_2^{\text{KL}} &= -\frac{1}{2} \log |\mathbf{C}| - \frac{1}{2} \text{tr}(\mathbf{C}^{-1})
 \end{aligned}$$

The following table shows the results of the regression analysis for the dependent variable "Number of children in the household" (N = 1,000). The independent variables are "Age of the head of household" and "Gender of the head of household". The results are presented in the following table:

1. The first step is to identify the problem. This involves understanding the current situation and the goals that need to be achieved.

The following table shows the number of persons employed in the various occupations in the city of New York, in 1880, and the number of persons employed in the same occupations in 1870. The occupations are classified into three groups: (1) Manufacturing, (2) Commerce, and (3) Services.

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Signed

Stephen Hordley

Dated

7 February 2003



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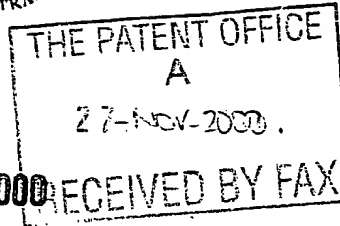


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1. Your reference

CM2478F

2. Patent application number

0028823.3

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27NOV00 E586865-1 000455

P01/7700 0.00-0028823.3

3. Full name, address and postcode of the or of each applicant (underline all surnames)

The Procter & Gamble Company
Cincinnati

Patents ADP number (if you know it)

06531859001

If the applicant is a corporate body, give the country/state of its incorporation

Ohio 45202, USA

4. Title of the invention

Cleaning methods and compositions

5. Name of your agent (if you have one)

"Address for service" in the United Kingdom to which all correspondence should be sent (including the postcode)

Maxim Courtney Brooks
Procter & Gamble Technical Centres Limited
Newcastle Technical Centre
P.O. Box Forest Hall No. 2
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NE12 9TS
ENGLAND

Patents ADP number (if you know it)

07582604001

6. If you are declaring priority from one or more earlier patent applications, give the country and the date of filing of the or of each of these earlier applications and (if you know it) the or each application number

Country

Priority application number
(if you know it)

Date of filing
(day / month / year)

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7. If this application is divided or otherwise derived from an earlier UK application, give the number and the filing date of the earlier application

Number of earlier application

Date of filing
(day / month / year)

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8. Is a statement of inventorship and of right to grant of a patent required in support of this request? (Answer 'Yes' if:

- a) any applicants named in part 3 is not an inventor, or
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18

Claim(s)

2

Abstract

1

Drawing(s)

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Statement of inventorship and right to grant of a patent (Patents Form 7/77)

Request for preliminary examination and search (Patents Form 9/77)

Request for substantive examination (Patents Form 10/77)

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11.

I/We request the grant of a patent on the basis of this application.

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*M C Brooks*Date 27 November 2000**M C Brooks Agent for the Applicant**

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DUPLICATE

CM2478F

CLEANING METHODS and COMPOSITIONS

Technical field

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The present invention is in the field of dishwashing, in particular it relates to dishwashing methods including methods for washing dishware/tableware in an automatic dishwashing machine using dishwashing products in multi-compartment pouch form. The methods of the invention provide excellent cleaning results.

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Background of the invention

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Unitised doses of dishwashing detergents are found to be more attractive and convenient to some consumers because they avoid the need of the consumer to measure the product thereby giving rise to a more precise dosing and avoiding wasteful overdosing or underdosing. For this reason automatic dishwashing detergent products in tablet form have become very popular. Detergent products in pouch form are also known in the art.

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It is well known to use bleach in dishwashing detergent formulations in order to remove stains, especially tea, coffee, fruit juice and carotenoid stains. Chlorine and peroxygen bleaches are effective for stain removal. While chlorine bleach is a very effective cleaning agent, it is not compatible with a variety of detergent ingredients and may require additional processing in order to be incorporated into a final product. Peroxide bleaches on the other hand are more compatible with conventional detergent ingredients. However, one of the problems found when formulating peroxygen containing dishwashing detergent compositions is the fact that the bleach is liable to decompose in contact with moisture, thereby reducing the amount of active bleach available for the dishwashing process. Once the decomposition process is initiated, moreover, decomposition is auto catalysed by the presence of free radicals generated by the decomposition process. The products of bleach decomposition can also oxidise detergency enzymes, thereby reducing the amount of enzyme available for the dishwashing process.

In the case of flexible unitised doses such as pouches, capsules or sachets which are moisture permeable, bleach decomposition gives rise to an additional problem due to the generation of gaseous oxygen. Usually the pouch material is not permeable to oxygen and this can lead to bloating or even destruction of the pouch and to a detrimental effect on appearance and on dispenser fit.

Some detergent ingredients used in dishwashing detergent compositions are liquids. These liquid ingredients can be difficult or costly to include in a solid detergent composition. Also, certain ingredients are preferably transported and supplied to detergent manufacturers in a liquid form and require additional, and sometimes costly, process steps to enable them to be included in a solid detergent composition. An example of these detergent ingredients are surfactants, especially nonionic surfactants which are typically liquid at room temperature or are typically transported and supplied to detergent manufacturers in liquid form. Another example are organic solvents.

Current methods of incorporating liquid ingredients into solid detergent compositions include absorbing the liquid ingredient onto a solid carrier, for example by mixing, agglomeration or spray-on techniques. Typically, solid detergent compositions comprise only low amounts of these liquid detergent ingredients due to the difficulty and expense of incorporating these liquid ingredients into a solid detergent. Furthermore, the incorporation of liquid ingredients into solid detergent compositions can impact on the dissolution characteristics of the composition (for example as the result of forming surfactant gel phases) and can also lead to problems of flowability. It would be advantageous to have a detergent composition which allows the different ingredients to be in their natural state i.e., liquid or solid. This would facilitate the manufacturing process and furthermore allow the delivery of liquid ingredients prior or post to the delivery of solid ingredients. For example differential dissolution of active ingredients would be beneficial in the case of enzyme/bleach compositions to avoid oxidation of enzymes by the bleach in the dishwashing liquor. It would also be advantageous to separate bleach from perfume.

An objective of the present invention is to provide dishwashing methods and products delivering improved cleaning performance and product stability. Another object is to provide

dishwashing methods and products which have simplified processing, which allow for the problems of product incompatibility and which are capable of providing differential dissolution of active components.

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Summary of the invention

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According to a first aspect of the present invention, there is provided a method of washing dishware/tableware in an automatic dishwashing machine, the method comprising simultaneously or sequentially delivering quantities of a particulate automatic dishwashing product and of an anhydrous liquid, gel or paste form dishwashing detergent auxiliary contained in separate compartments of a multi-compartment pouch into the same or different cycles of the dishwashing machine.

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Suitable multi-compartment pouches for use herein include water-soluble, water-dispersible and water-permeable pouches. Preferred for use herein are water soluble pouches, based on partially hydrolysed polyvinyl alcohol as pouch material. Although, soluble in water, these pouches have the disadvantage that they are permeable to moisture.

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The term anhydrous as used herein is intended to include compositions containing less than about 10% of water by weight of the composition, preferably less than about 5% of water and more preferably less than about 1%. The water can be present in the form of hydrated compounds, i.e. bound water or in the form of moisture. It is preferred that the composition contains less than about 1%, preferably less than about 0.1% free moisture. Free moisture can be measured by extracting 2 g of the product into 50 ml of dry methanol at room temperature for 20 minutes and then analysis a 1 ml aliquot of the methanol by Karl Fischer titration.

30

According to a preferred embodiment of the present invention, the particulate dishwashing product comprises one or more moisture-sensitive detergent actives and the detergent auxiliary comprises a humectant in levels sufficient to act as a moisture sink for stabilising the moisture-sensitive detergent active. A detergent active is considered to be moisture-sensitive when it can interact with moisture to decrease its detergency activity as for example

detergency bleach. Particulate bleaches suitable for use herein include inorganic peroxides inclusive of perborates and percarbonates, organic peracids inclusive of preformed monoperoxy carboxylic acids, such as phthaloyl amido peroxy hexanoic acid and di-acyl peroxides. Preferred peroxides for use herein are percarbonate and perborate bleach.

5 Humectant is a substance which can pick up or emit moisture to the surroundings depending on the surrounding relative humidity. When formulated as part of the detergent auxiliary, the humectants used herein are capable of acting as moisture sink for the powder layer. This stabilises the moisture-sensitive detergent active. The humectants should have a humidity
10 equilibrium point such as to enable them to act as moisture sink but preferably they should take up less than about 10%, more preferably less than about 5% even more preferably less than about 1% of water at a relative humidity of 50% or less, preferably at relative humidity of 75% or less, and more preferably at relative humidity of 90% or less under ambient conditions of temperature and pressure (20°C and 1 atmosphere). Humectants suitable for
15 use herein include non-aqueous hydrophilic organic solvents inclusive of glycols and polyhydric alcohols, for example sorbitol, glycerol, dipropylene glycol and mixtures thereof, and also various hygroscopic solids inclusive of inorganic or organic salts such as silicates, phosphates and citrates, as well as sugars, etc.

20 In a preferred embodiment, the anhydrous detergent auxiliary composition comprises a detergency enzyme. The enzyme is preferably in liquid form and is delivered to the wash liquor substantially prior to the particulate products, thereby minimizing or avoiding interaction with actives, such as bleach, which can have a deleterious effect on enzyme stability and performance in the wash solution.

25 In preferred embodiments the dishwashing composition comprises an organic solvent system. The organic solvent system can simply act as a liquid carrier, but in preferred compositions, the solvent can aid removal of cooked-, baked- or burnt-on soil and thus has detergent functionality in its own right. The organic solvent system (comprising a single solvent
30 compound or a mixture of solvent compounds) preferably has a volatile organic content above 1 mm Hg and more preferably above 0.1 mm Hg of less than about 50%, preferably less than about 20% and more preferably less than about 10% by weight of the solvent

system. Herein volatile organic content of the solvent system is defined as the content of organic components in the solvent system having a vapor pressure higher than the prescribed limit at 25°C and atmospheric pressure.

- 5 The organic solvent system for use herein is preferably selected from organoamine solvents, inclusive of alkanolamines, alkylamines, alkyleneamines and mixtures thereof; alcoholic solvents inclusive of aromatic, aliphatic (preferably C₄-C₁₀) and cycloaliphatic alcohols and mixtures thereof; glycols and glycol derivatives inclusive of C₂-C₃ (poly)alkylene glycols, glycol ethers, glycol esters and mixtures thereof; and mixtures selected from organoamine
- 10 solvents, alcoholic solvents, glycols and glycol derivatives. In one preferred embodiment the organic solvent comprises organoamine (especially alkanolamine) solvent and glycol ether solvent, preferably in a weight ratio of from about 3:1 to about 1:3, and wherein the glycol ether solvent is selected from ethylene glycol monobutyl ether, diethylene glycol monobutyl ether, ethylene glycol monomethyl ether, ethylene glycol monoethyl ether, diethylene glycol
- 15 monomethyl ether, diethylene glycol monoethyl ether, propylene glycol monobutyl ether, and mixtures thereof. Preferably, the glycol ether is a mixture of diethylene glycol monobutyl ether and propylene glycol butyl ether, especially in a weight ratio of from about 1:2 to about 2:1.
- 20 According to another embodiment of the invention the dishwashing auxiliary composition can take the form of a paste having a density greater than about 1100 Kg/m³, preferably greater than about 1300 Kg/m³.

Multi-compartment pouches suitable for use herein can include compartments with different

25 solubility profiles controlled by for example pH, temperature or any other means. High temperature water-soluble pouches allow handling of the pouches at ambient temperature with wet hands.

The multi-compartment pouches herein comprise at least one compartment containing a

30 powder composition and at least one containing an anhydrous liquid, gel or paste. This powder composition comprises traditional solid materials used in dishwashing detergent, such as builders, alkalinity sources, together with moisture-sensitive detergent active such as

bleaches, etc. The liquid, gel or paste compositions comprise traditional liquid materials used in dishwashing detergents, such as non-ionic surfactants or the organic solvents described hereinabove together with a humectant. Preferably the compartment comprising the detergent auxiliary is placed above the compartment comprising the moisture-sensitive detergent active in order to help protect the moisture-sensitive detergent active and to reduce the surface area of the pouch containing compartment which is exposed to moisture.

Detailed description of the invention

The present invention envisages the use of dishwashing detergent and auxiliary compositions in a multi-compartment pouch, whereby a moisture-sensitive detergent active and a humectant are contained in separate compartments. The humectant is capable of acting as a moisture sink and acts to stabilize the moisture-sensitive detergent active.

Unitised dose forms specially useful for use herein are pouches. The pouch herein is typically a closed structure which comprises two or more compartments, made of materials described herein. Subject to the constraints of dispenser fit, the pouch can be of any form, shape and material which is suitable to hold the composition, e.g. without allowing the release of the composition from the pouch prior to contact of the pouch to water. The exact execution will depend on, for example, the type and amount of the composition in the pouch, the number of compartments in the pouch, the characteristics required from the pouch to hold, protect and deliver or release the composition and/or components thereof.

The composition, or components thereof, are contained in the internal volume space of the pouch, and are typically separated from the outside environment by a barrier of water-soluble material. Typically, different components of the composition contained in different compartments of the pouch are separated from one another by a barrier of water-soluble material.

In the case of multi-compartment pouches, the compartments may be of a different colour from each other, for example a first compartment may be green or blue, and a second compartment may be white or yellow. One compartment of the pouch may be opaque or

semi-opaque, and a second compartment of the pouch may be translucent, transparent, or semi-transparent. The compartments of the pouch may be the same size, having the same internal volume, or may be different sizes having different internal volumes.

- 5 For reasons of deformability and dispenser fit under compression forces, pouches or pouch compartments containing a component which is liquid will usually contain an air bubble having a volume of up to about 50%, preferably up to about 40%, more preferably up to about 30%, more preferably up to about 20%, more preferably up to about 10% of the volume space of said compartment.

10 The pouch is preferably made of a pouch material which is soluble or dispersible in water, and has a water-solubility of at least 50%, preferably at least 75% or even at least 95%, as measured by the method set out here after using a glass-filter with a maximum pore size of 20 microns.

15 50 grams \pm 0.1 gram of pouch material is added in a pre-weighed 400 ml beaker and 245ml \pm 1ml of distilled water is added. This is stirred vigorously on a magnetic stirrer set at 600 rpm, for 30 minutes. Then, the mixture is filtered through a folded qualitative sintered-glass filter with a pore size as defined above (max: 20 micron). The water is dried off from the collected
20 filtrate by any conventional method, and the weight of the remaining material is determined (which is the dissolved or dispersed fraction). Then, the % solubility or dispersability can be calculated.

25 Preferred pouch materials are polymeric materials, preferably polymers which are formed into a film or sheet. The pouch material can, for example, be obtained by casting, blow-moulding, extrusion or blow extrusion of the polymeric material, as known in the art.

30 Preferred polymers, copolymers or derivatives thereof suitable for use as pouch material are selected from polyvinyl alcohols, polyvinyl pyrrolidone, polyalkylene oxides, acrylamide, acrylic acid, cellulose, cellulose ethers, cellulose esters, cellulose amides, polyvinyl acetates, polycarboxylic acids and salts, polyaminoacids or peptides, polyamides, polyacrylamide, copolymers of maleic/acrylic acids, polysaccharides including starch and gelatine, natural

gums such as xanthum and carragum. More preferred polymers are selected from polyacrylates and water-soluble acrylate copolymers, methylcellulose, carboxymethylcellulose sodium, dextrin, ethylcellulose, hydroxyethyl cellulose, hydroxypropyl methylcellulose, maltodextrin, polymethacrylates, and most preferably selected from polyvinyl alcohols, polyvinyl alcohol copolymers and hydroxypropyl methyl cellulose (HPMC), and combinations thereof. Preferably, the level of polymer in the pouch material, for example a PVA polymer, is at least 60%.

The polymer can have any weight average molecular weight, preferably from about 1000 to 1,000,000, more preferably from about 10,000 to 300,000 yet more preferably from about 20,000 to 150,000.

Mixtures of polymers can also be used as the pouch material. This can be beneficial to control the mechanical and/or dissolution properties of the compartments or pouch, depending on the application thereof and the required needs. Suitable mixtures include for example mixtures wherein one polymer has a higher water-solubility than another polymer, and/or one polymer has a higher mechanical strength than another polymer. Also suitable are mixtures of polymers having different weight average molecular weights, for example a mixture of PVA or a copolymer thereof of a weight average molecular weight of about 10,000- 40,000, preferably around 20,000, and of PVA or copolymer thereof, with a weight average molecular weight of about 100,000 to 300,000, preferably around 150,000.

Also suitable herein are polymer blend compositions, for example comprising hydrolytically degradable and water-soluble polymer blends such as polylactide and polyvinyl alcohol, obtained by mixing polylactide and polyvinyl alcohol, typically comprising about 1-35% by weight polylactide and about 65% to 99% by weight polyvinyl alcohol.

Preferred for use herein are polymers which are from about 60% to about 98% hydrolysed, preferably about 80% to about 90% hydrolysed, to improve the dissolution characteristics of the material.

Most preferred pouch materials are PVA films known under the trade reference Monosol M8630, as sold by Chris-Craft Industrial Products of Gary, Indiana, US, and PVA films of corresponding solubility and deformability characteristics. Other films suitable for use herein include films known under the trade reference PT film or the K-series of films supplied by Aicello, or VF-HP film supplied by Kuraray.

The pouch material herein can also comprise one or more additive ingredients. For example, it can be beneficial to add plasticisers, for example glyccrol, ethylene glycol, diethyleneglycol, propylene glycol, sorbitol and mixtures thereof. Other additives include functional detergent additives to be delivered to the wash water, for example organic polymeric dispersants, etc.

The pouch can be prepared according to methods known in the art. The pouch is typically prepared by first cutting an appropriately sized piece of pouch material, preferably the pouch material. The pouch material is then folded to form the necessary number and size of compartments and the edges are sealed using any suitable technology, for example heat sealing, wet sealing or pressure sealing. Preferably, a sealing source is brought into contact with the pouch material, heat or pressure is applied and the pouch material is sealed.

The pouch material is typically introduced to a mould and a vacuum applied so that the pouch material is flush with the inner surface of the mould, thus forming a vacuum formed indent or niche in said pouch material. This is referred to as vacuum-forming.

Another suitable method is thermo-forming. Thermo-forming typically involves the step of forming an open pouch in a mould under application of heat, which allows the pouch material to take on the shape of the mould.

Typically more than one piece of pouch material is used for making multi-compartment pouches. For example, a first piece of pouch material can be vacuum pulled into the mould so that said pouch material is flush with the inner walls of the mould. A second piece of pouch material can then be positioned such that it at least partially overlaps, and preferably completely overlaps, with the first piece of pouch material. The first piece of pouch material

and second piece of pouch material are sealed together. The first piece of pouch material and second piece of pouch material can be made of the same type of material or can be different types of material.

5 In a preferred process, a piece of pouch material is folded at least twice, or at least three pieces of pouch material are used, or at least two pieces of pouch material are used wherein at least one piece of pouch material is folded at least once. The third piece of pouch material, or a folded piece of pouch material, creates a barrier layer that, when the sachet is sealed, divides the internal volume of said sachet into at least two or more compartments.

10 The pouch can also be prepared by fitting a first piece of the pouch material into a mould, for example the first piece of film may be vacuum pulled into the mould so that said film is flush with the inner walls of the mould. A composition, or component thereof, is typically poured into the mould. A pre-sealed compartment made of pouch material, is then typically placed
15 over the mould containing the composition, or component thereof. The pre-sealed compartment preferably contains a composition, or component thereof. The pre-sealed compartment and said first piece of pouch material may be sealed together to form the pouch.

The detergent auxiliary herein can comprise traditional detergency components and can also
20 comprise organic solvents having a cleaning function and organic solvents having a carrier or diluent function or some other specialised function. The compositions will generally be built and comprise one or more detergent active components which may be selected from colorants, bleaching agents, surfactants, alkalinity sources, enzymes, thickeners (in the case of liquid, paste, cream or gel compositions), anti-corrosion agents (e.g. sodium silicate),
25 hydrotropes (e.g. sodium cumene sulfate) and disrupting and binding agents (in the case of powder, granules or tablets). Highly preferred detergent auxiliary components include a builder compound, an alkalinity source, a surfactant, an enzyme and a bleaching agent.

Unless otherwise specified, the components described hereinbelow can be incorporated either
30 in the automatic dishwashing product or detergent auxiliary.

The organic solvents should be selected so as to be compatible with the tableware/cookware as well as with the different parts of an automatic dishwashing machine. Furthermore, the solvent system should be effective and safe to use having a volatile organic content above 1 mm Hg (and preferably above 0.1 mm Hg) of less than about 50%, preferably less than about 30%, more preferably less than about 10% by weight of the solvent system. Also they should have very mild pleasant odours. The individual organic solvents used herein generally have a boiling point above about 150°C, flash point above about 100°C and vapor pressure below about 1 mm Hg, preferably below 0.1 mm Hg at 25°C and atmospheric pressure.

10 Solvents that can be used herein include: i) alcohols, such as benzyl alcohol, 1,4-cyclohexanedimethanol, 2-ethyl-1-hexanol, furfuryl alcohol, 1,2-hexanediol and other similar materials; ii) amines, such as alkanolamines (e.g. primary alkanolamines: monoethanolamine, monoisopropanolamine, diethylethanolamine, ethyl diethanolamine; secondary alkanolamines: diethanolamine, diisopropanolamine, 2-(methylamino)ethanol; ternary alkanolamines: triethanolamine, triisopropanolamine); alkylamines (e.g. primary alkylamines: monomethylamine, monoethylamine, monopropylamine, monobutylamine, monopentylamine, cyclohexylamine), secondary alkylamines: (dimethylamine), alkylene amines (primary alkylene amines: ethylenediamine, propylenediamine) and other similar materials; iii) esters, such as ethyl lactate, methyl ester, ethyl acetoacetate, ethylene glycol monobutyl ether acetate, diethylene glycol monoethyl ether acetate, diethylene glycol monobutyl ether acetate and other similar materials; iv) glycol ethers, such as ethylene glycol monobutyl ether, diethylene glycol monobutyl ether, ethylene glycol monomethyl ether, ethylene glycol monoethyl ether, diethylene glycol monomethyl ether, diethylene glycol monoethyl ether, propylene glycol butyl ether and other similar materials; v) glycols, such as propylene glycol, diethylene glycol, hexylene glycol (2-methyl-2, 4 pentanediol), triethylene glycol, composition and dipropylene glycol and other similar materials; and mixtures thereof.

Surfactant

30 In the methods of the present invention, the detergent surfactant is preferably low foaming by itself or in combination with other components (i.e. suds suppressers). Surfactants suitable herein include anionic surfactants such as alkyl sulfates, alkyl ether sulfates, alkyl benzene sulfonates, alkyl glyceryl sulfonates, alkyl and alkenyl sulphonates, alkyl ethoxy

carboxylates, N-acyl sarcosinates, N-acyl taurates and alkyl succinates and sulfosuccinates, wherein the alkyl, alkenyl or acyl moiety is C₅-C₂₀, preferably C₁₀-C₁₈, linear or branched; cationic surfactants such as chlorine esters (US-A-4228042, US-A-4239660 and US-A-4260529) and mono C₆-C₁₆ N-alkyl or alkenyl ammonium surfactants wherein the remaining N positions are substituted by methyl, hydroxyethyl or hydroxypropyl groups; low and high cloud point nonionic surfactants and mixtures thereof including nonionic alkoxyated surfactants (especially ethoxylates derived from C₆-C₁₈ primary alcohols), ethoxylated-propoxylated alcohols (e.g., Olin Corporation's Poly-Tergent® SLF18), epoxy-capped poly(oxyalkylated) alcohols (e.g., Olin Corporation's Poly-Tergent® SLF18B; see WO-A-94/22800), ether-capped poly(oxyalkylated) alcohol surfactants, and block polyoxyethylene-polyoxypropylene polymeric compounds such as PLURONIC®, REVERSED PLURONIC®, and TETRONIC® by the BASF-Wyandotte Corp., Wyandotte, Michigan; amphoteric surfactants such as the C₁₂-C₂₀ alkyl amine oxides (preferred amine oxides for use herein include lauryldimethyl amine oxide and hexadecyl dimethyl amine oxide), and alkyl amphocarboxylic surfactants such as Miranol™ C2M; and zwitterionic surfactants such as the betaines and sultaines; and mixtures thereof. Surfactants suitable herein are disclosed, for example, in US-A-3,929,678, US-A-4,259,217, EP-A-0414 549, WO-A-93/08876 and WO-A-93/08874. Surfactants are typically present at a level of from about 0.2% to about 30% by weight, more preferably from about 0.5% to about 10% by weight, most preferably from about 1% to about 5% by weight of composition. Preferred surfactant for use herein are low foaming and include low cloud point nonionic surfactants and mixtures of higher foaming surfactants with low cloud point nonionic surfactants which act as suds suppresser therefor.

Builder

Builders suitable for use in detergent and cleaning compositions herein include water-soluble builders such as citrates, carbonates and polyphosphates e.g. sodium tripolyphosphate and sodium tripolyphosphate hexahydrate, potassium tripolyphosphate and mixed sodium and potassium tripolyphosphate salts; and partially water-soluble or insoluble builders such as crystalline layered silicates (EP-A-0164514 and EP-A-0293640) and aluminosilicates inclusive of Zeolites A, B, P, X, HS and MAP. The builder is typically present at a level of

from about 1% to about 80% by weight, preferably from about 10% to about 70% by weight, most preferably from about 20% to about 60% by weight of composition.

Amorphous sodium silicates having an $\text{SiO}_2:\text{Na}_2\text{O}$ ratio of from 1.8 to 3.0, preferably from 1.8 to 2.4, most preferably 2.0 can also be used herein although highly preferred from the viewpoint of long term storage stability are compositions containing less than about 22%, preferably less than about 15% total (amorphous and crystalline) silicate.

Enzyme

Enzymes suitable herein include bacterial and fungal cellulases such as Carczyme and Celluzyme (Novo Nordisk A/S); peroxidases; lipases such as Amano-P (Amano Pharmaceutical Co.), M1 Lipase^R and Lipomax^R (Gist-Brocades) and Lipolase^R and Lipolase Ultra^R (Novo); cutinases; proteases such as Esperase^R, Alcalase^R, Durazym^R and Savinase^R (Novo) and Maxatase^R, Maxacal^R, Properase^R and Maxapem^R (Gist-Brocades); and α and β amylases such as Purafect-Ox Am^R (Genencor) and Termamyl^R, Ban^R, Fungamyl^R, Duramyl^R, and Natalase^R (Novo); and mixtures thereof. Enzymes are preferably added herein as prills, granulates, or cogranulates at levels typically in the range from about 0.0001% to about 2% pure enzyme by weight of composition.

Bleaching agent

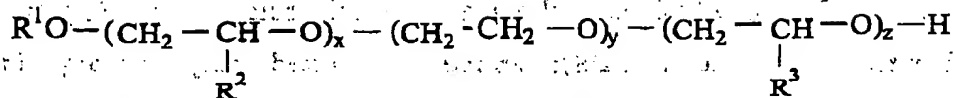
Bleaching agents suitable herein include chlorine and oxygen bleaches, especially inorganic perhydrate salts such as sodium perborate mono- and tetrahydrates and sodium percarbonate optionally coated to provide controlled rate of release (see, for example, GB-A-1466799 on sulfate/carbonate coatings), preformed organic peroxyacids and mixtures thereof with organic peroxyacid bleach precursors and/or transition metal-containing bleach catalysts (especially manganese or cobalt). Inorganic perhydrate salts are typically incorporated at levels in the range from about 1% to about 40% by weight, preferably from about 2% to about 30% by weight and more preferably from about 5% to about 25% by weight of composition. Peroxyacid bleach precursors preferred for use herein include precursors of perbenzoic acid and substituted perbenzoic acid; cationic peroxyacid precursors; peracetic acid precursors

such as TAED, sodium acetoxybenzene sulfonate and pentaacetylglucose; pernonanoic acid precursors such as sodium 3,5,5-trimethylhexanoyloxybenzene sulfonate (iso-NOBS) and sodium nonanoyloxybenzene sulfonate (NOBS); amide substituted alkyl peroxyacid precursors (EP-A-0170386); and benzoxazin peroxyacid precursors (EP-A-0332294 and EP-A-0482807). Bleach precursors are typically incorporated at levels in the range from about 0.5% to about 25%, preferably from about 1% to about 10% by weight of composition while the preformed organic peroxyacids themselves are typically incorporated at levels in the range from 0.5% to 25% by weight, more preferably from 1% to 10% by weight of composition. Bleach catalysts preferred for use herein include the manganese triazacyclononane and related complexes (US-A-4246612, US-A-5227084); Co, Cu, Mn and Fe bispyridylamine and related complexes (US-A-5114611); and pentamine acetate cobalt(III) and related complexes(US-A-4810410).

Low cloud point non-ionic surfactants and suds suppressers

The suds suppressers suitable for use herein include nonionic surfactants having a low cloud point. "Cloud point", as used herein, is a well known property of nonionic surfactants which is the result of the surfactant becoming less soluble with increasing temperature, the temperature at which the appearance of a second phase is observable is referred to as the "cloud point" (See Kirk Othmer, pp. 360-362). As used herein, a "low cloud point" nonionic surfactant is defined as a nonionic surfactant system ingredient having a cloud point of less than 30° C., preferably less than about 20° C., and even more preferably less than about 10° C., and most preferably less than about 7.5° C. Typical low cloud point nonionic surfactants include nonionic alkoxyated surfactants, especially ethoxylates derived from primary alcohol, and polyoxypropylene/polyoxyethylene/polyoxypropylene (PO/EO/PO) reverse block polymers. Also, such low cloud point nonionic surfactants include, for example, ethoxylated-propoxylated alcohol (e.g., Olin Corporation's Poly-Tergent® SLF18) and epoxy-capped poly(oxyalkylated) alcohols (e.g., Olin Corporation's Poly-Tergent® SLF18B series of nonionics, as described, for example, in US-A-5,576,281).

Preferred low cloud point surfactants are the ether-capped poly(oxyalkylated) suds suppresser having the formula:



wherein R¹ is a linear, alkyl hydrocarbon having an average of from about 7 to about 12 carbon atoms, R² is a linear, alkyl hydrocarbon of about 1 to about 4 carbon atoms, R³ is a linear, alkyl hydrocarbon of about 1 to about 4 carbon atoms, x is an integer of about 1 to about 6, y is an integer of about 4 to about 15, and z is an integer of about 4 to about 25.

Other low cloud point nonionic surfactants are the ether-capped poly(oxyalkylated) having the formula:



wherein, R_I is selected from the group consisting of linear or branched, saturated or unsaturated, substituted or unsubstituted, aliphatic or aromatic hydrocarbon radicals having from about 7 to about 12 carbon atoms; R_{II} may be the same or different, and is independently selected from the group consisting of branched or linear C₂ to C₇ alkylene in any given molecule; n is a number from 1 to about 30; and R_{III} is selected from the group consisting of:

- (i) a 4 to 8 membered substituted, or unsubstituted heterocyclic ring containing from 1 to 3 hetero atoms; and
- (ii) linear or branched, saturated or unsaturated, substituted or unsubstituted, cyclic or acyclic, aliphatic or aromatic hydrocarbon radicals having from about 1 to about 30 carbon atoms;
- (b) provided that when R² is (ii) then either: (A) at least one of R¹ is other than C₂ to C₃ alkylene; or (B) R³ has from 6 to 30 carbon atoms, and with the further proviso that when R² has from 8 to 18 carbon atoms, R is other than C₁ to C₃ alkyl.

Other suitable components herein include organic polymers having dispersant, anti-redeposition, soil release or other detergency properties in levels of from about 0.1% to about 30%, preferably from about 0.5% to about 15%, most preferably from about 1% to about 10% by weight of composition. Preferred anti-redeposition polymers herein include acrylic acid containing polymers such as Sokalan PA30, PA20, PA15, PA10 and

Sokalan CP10 (BASF GmbH), Acusol 45N, 480N, 460N (Rohm and Haas), acrylic acid/maleic acid copolymers such as Sokalan CP5 and acrylic/methacrylic copolymers. Preferred soil release polymers herein include alkyl and hydroxyalkyl celluloses (US-A-4,000,093), polyoxyethylenes, polyoxypropylenes and copolymers thereof, and nonionic and
 5 anionic polymers based on terephthalate esters of ethylene glycol, propylene glycol and mixtures thereof.

Heavy metal sequestrants and crystal growth inhibitors are suitable for use herein in levels generally from about 0.005% to about 20%, preferably from about 0.1% to about 10%, more
 10 preferably from about 0.25% to about 7.5% and most preferably from about 0.5% to about 5% by weight of composition, for example diethylenetriamine penta (methylene phosphonate), ethylenediamine tetra(methylene phosphonate) hexamethylenediamine tetra(methylene phosphonate), ethylene diphosphonate, hydroxy-ethylene-1,1-diphosphonate, nitrilotriacetate, ethylenediaminetetracetate, ethylenediamine-N,N'-disuccinate in their salt
 15 and free acid forms.

The compositions herein can contain a corrosion inhibitor such as organic silver coating agents in levels of from about 0.05% to about 10%, preferably from about 0.1% to about 5% by weight of composition (especially paraffins such as Winog 70 sold by Wintershall,
 20 Salzbergen, Germany), nitrogen-containing corrosion inhibitor compounds (for example benzotriazole and benzimidazole - see GB-A-1137741) and Mn(II) compounds, particularly Mn(II) salts of organic ligands in levels of from about 0.005% to about 5%, preferably from about 0.01% to about 1%, more preferably from about 0.02% to about 0.4% by weight of the composition.

Other suitable components herein include colorants, water-soluble bismuth compounds such as bismuth acetate and bismuth citrate at levels of from about 0.01% to about 5%, enzyme stabilizers such as calcium ion, boric acid, propylene glycol and chlorine bleach scavengers at levels of from about 0.01% to about 6%, lime soap dispersants (see WO-A-93/08877), suds
 30 suppressors (see WO-93/08876 and EP-A-0705324), polymeric dye transfer inhibiting agents, optical brighteners, perfumes, fillers and clay.

Liquid detergent compositions can contain low quantities of low molecular weight primary or secondary alcohols such as methanol, ethanol, propanol and isopropanol can be used in the liquid detergent of the present invention. Other suitable carrier solvents used in low quantities includes glycerol, propylene glycol, ethylene glycol, 1,2-propanediol, sorbitol and mixtures thereof.

Examples

Abbreviations used in Examples

10 In the examples, the abbreviated component identifications have the following meanings:

Carbonate	: Anhydrous sodium carbonate
STPP (anhydrous)	: Sodium tripolyphosphate anhydrous
STPP (hydrated)	: Sodium tripolyphosphate hydrated to approximately 8%
Silicate	: Amorphous Sodium Silicate ($\text{SiO}_2:\text{Na}_2\text{O}$ = from 2:1 to 4:1)
HEDP	: Ethane 1-hydroxy-1,1-diphosphonic acid
Perborate	: Sodium perborate monohydrate
Percarbonate	: Sodium percarbonate of the nominal formula $2\text{Na}_2\text{CO}_3 \cdot 3\text{H}_2\text{O}_2$
Termamyl	: α -amylase available from Novo Nordisk A/S
Savinase	: protease available from Novo Nordisk A/S
FN3	: protease available from Genencor
SLF18	: low foaming surfactant available from Olin Corporation
ACNI	: alkyl capped non-ionic surfactant of formula $\text{C}_{9/11} \text{H}_{19/23} \text{EO}_8$ -cyclohexyl acetal
C_{14}AO	: tetradecyl dimethyl amine oxide
C_{16}AO	: hexadecyl dimethyl amine oxide
Duramyl	: α -amylase available from Novo Nordisk A/S
DPG	: dipropylene glycol

In the following examples all levels are quoted as parts by weight.

Examples 1 to 4

The compositions of examples 1 to 4 are made in the form of a two compartment PVA pouch. The dual compartment pouch is made from a Monosol M8630 film as supplied by Chris-Craft Industrial Products. The pouches made by presealing the liquid composition using the technique described hereinabove. The particulate composition and the anhydrous composition are placed in two different horizontal by-layered compartments, the anhydrous composition being placed above the particulate composition. The exemplified pouches show a good stability of the particulate automatic dishwashing product.

Example	1	2	3	4
<u>Particulate composition</u>				
C ₁₄ AO	5.55	5.55		
C ₁₆ AO			5.55	5.55
ACNI	5.55	5.55		
SLF18			5.55	5.55
STPP (anhydrous)	21.0	21.0	21.0	21.0
STPP (hydrated)	31.5	31.5	31.5	31.5
HEDP	1.0	1.0	1.0	1.0
Savinase	0.7	0.7	0.7	0.7
Termamyl	0.7	0.7	0.7	0.7
Perborate		13.55		13.55
Percarbonate	13.55		13.55	
Carbonate	15.0	10.0	15.0	15.0
Silicate	5.0	10.0	5.0	5.0
Perfume	0.5	0.5	0.5	0.5
<u>Anhydrous composition</u>				
DPG	99.5	95.0	95.0	99.5
FN3 Liquid		2.60	2.4	
Duramyl Liquid		2.0	2.4	
Dyc	0.5	0.4	0.2	0.5

Claims

1. A method of washing dishware/tableware in an automatic dishwashing machine, the method comprising simultaneously or sequentially delivering quantities of a particulate automatic dishwashing product and of an anhydrous liquid, gel or paste form dishwashing detergent auxiliary contained in separate compartments of a multi-compartment pouch into the same or different cycles of the dishwashing machine.

2. A method according to claim 1 wherein the particulate dishwashing product comprises one or more moisture-sensitive detergent actives and wherein the detergent auxiliary comprises a humectant in levels sufficient to act as a moisture sink for stabilising the moisture-sensitive detergent active.

3. A method according to claims 1 or 2 wherein the humectant is selected from non-aqueous hydrophilic organic solvent inclusive of glycols and polyhydric alcohols, for example sorbitol, glycerol, dipropylene glycol and mixtures thereof.

4. A method according to claims 2 or 3 wherein the moisture-sensitive detergent active is a detergency bleach.

5. A method according to any preceding claims wherein the anhydrous detergent auxiliary comprises a detergency enzyme.

6. A method according to the preceding claims wherein the anhydrous detergent auxiliary comprises an organic solvent system effective in removing cooked-, baked- and burnt-on soils.

7. A method according to claim 6 wherein the organic solvent system is selected from alcohols, amines, esters, glycol ethers, glycols, terpenes and mixtures thereof.

8. A method according to claims 6 or 7 wherein the organic solvent system is selected from organoamine solvents, inclusive of alkanolamines, alkylamines, alkyleneamines and

mixtures thereof; alcoholic solvents inclusive of aromatic, aliphatic (preferably C_4-C_{10}) and cycloaliphatic alcohols and mixtures thereof; glycols and glycol derivatives inclusive of C_2-C_{10} (poly)alkylene glycols, glycol ethers, glycol esters and mixtures thereof; and mixtures selected from organoamine solvents, alcoholic solvents, glycols and glycol derivatives.

9. A method according to any of claims 6 to 8 wherein the organic solvent comprises organoamine (especially alkanolamine) solvent and glycol ether solvent, preferably in a weight ratio of from about 3:1 to about 1:3, and wherein the glycol ether solvent is selected from ethylene glycol monobutyl ether, diethylene glycol monobutyl ether, ethylene glycol monomethyl ether, ethylene glycol monoethyl ether, diethylene glycol monomethyl ether, diethylene glycol monoethyl ether, propylene glycol monobutyl ether, dipropylene glycol monobutyl ether, ethylene glycol phenyl ether and mixtures thereof.

10. A method according to any preceding claim wherein the detergent auxiliary is in the form of a paste having a density greater than about 1100 Kg/m^3 , preferably greater than about 1300 Kg/m^3 .

11. A method according to any preceding claim wherein the pouch is water soluble.

12. A method according to claim 11 wherein the compartments of the multi-compartment pouch have different rates of solubility in water under given temperature conditions.

13. A method according to any preceding claim wherein the anhydrous detergent auxiliary composition comprises a non-ionic surfactant.

ABSTRACT

A method of washing dishware/tableware in an automatic dishwashing machine, the method comprising simultaneously or sequentially delivering quantities of a particulate automatic dishwashing product and of an anhydrous liquid, gel or paste form dishwashing detergent auxiliary contained in separate compartments of a multi-compartment pouch into the same or different cycles of the dishwashing machine. The method provides improved cleaning performance and product stability.

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